DEFLUORIDATION OF POTABLE WATER IN CKDu PREVALENT AREAS ENRICHED WITH HARDNESS USING MODIFIED-FLY ASH FUNCTIONALIZED WITH IRON OXIDE

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Thesis submitted in partial fulfillment of the requirements for the degree of Master of Environmental Management

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September 2020

DECLARATION

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Abstract

Chronic Kidney Disease of Unknown Etiology is a major health issue reported in countries around the equator including Sri Lanka, Tunisia, Andra Pradesh India, and El Salvador. In Sri Lanka, CKDu is prevalent in the North-Central Province and now it is being progressed in the dry zone. The exact causal factor for the disease is not known yet where scientists now mainly suspect that the multiple ion interaction in potable groundwater may be the root cause for the disease. Further, fluoride ion concentration is higher in CKDu prevalent areas and the interaction of fluoride ion with other constituents (Cd, As, Al, hardness) is mainly suspected as the cause for the disease. The synergistic effect of hardness and fluoride on the CKDu had been discussed in many studies worldwide where the prevalence of other ions is very less, and it is below the WHO maximum allowed concentrations. The hardness and fluoride distribution are relatively higher in CKDu prevalent areas in Sri Lanka and the nephrotoxicity of hardness and fluoride in their mutual presence is proven by experiments with mice. Our preliminary studies found that there is no CKDu when the hardness and fluoride concentrations of water are below 200.00 mg/L and 0.47 mg/L, respectively. Therefore, this study was carried out to remove the hardness and fluoride concentrations of water to the level of 200.00mg/L and 0.47mg/L, respectively. Initially, a divalent cation exchange column was designed using a commercially available cation exchange resin, ECO A, to remove excessive hardness level up to 200.00mg/L. The eluent from the cation exchange column was further treated for defluoridation. Coal derived fly ash was further modified using the hydrothermal method. The Modified Fly Ash (MFA) was further treated with Fe (III) Chloride to generate positive charges on the surface. FTIR, SEM, EDX, confirmed the incorporation of Fe into MFA and, the defluoridation ability of Fe functionalized MFA. FTIR spectra (400cm⁻¹ – 600cm⁻¹ region) showed the incorporation of Fe into MFA. The average crystalline size obtained from XRD analysis was 23.3nm and the synthesized material was in nanoscale. The batch experiments showed that 1.3g of the Fe functionalized MFA resulted in the maximum defluoridation for a 100.00 ml of water sample containing 200.00 mg/L hardness and 2.00 mg/L fluoride within 40 minutes of contact time at pH 6. The material gave optimum defluoridation at pH 6 and therefore there is no need of altering the pH of water for the defluoridation. The adsorption data fitted with the Langmuir adsorption isotherm where the maximum adsorption capacity was 10.00mg/g. The separation factor for the Langmuir adsorption was 1.23 and therefore the Langmuir adsorption is favorable. The reaction followed pseudo second order kinetics. Regeneration studies of the Fe functionalized MFA showed that NaOH was the best regeneration agent and the material was exhausted after two regeneration cycles. The material synthesize cost to purify water for a five-member family for three months was LKR 6923.07 (37.52USD) and the cost for the regeneration was LKR 174.46 (0.95USD). Therefore, the synthesized material is ideal and cost effective to remove fluoride in potable ground water in CKDu prevalent areas.

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Key Words: CKDu, Functionalized Modified Fly Ash (FMFA), Defluoridation, Fluoride, Hardness, Adsorption, Iron Oxide

ACKNOWLEDGEMENT

First and foremost, I would like to extend my heartfelt gratitude to my supervisor Prof. Mahesh. Jayaweera, for providing me the opportunity to complete the research. Your guidance and encouragement given at every step of the way in the research helped me to achieve the goals. Your support was immense, and I am very fortunate to have you as my supervisor. I extremely appreciate the advice given in experiments, writings, and moral assistance given to complete this research.

I am very grateful to my co-supervisor Dr. Buddhika Gunawardana, for providing her guidance and support for the research project. Your feedback on my experiments, writing, helped me to complete a very productive study. Furthermore, I would like to extend my gratitude to Prof. Jagath Manatunge for guiding me to conduct a fruitful and successful research. Your advice and guidance helped me to think out of the box.

I wish to express my sincere thanks to the laboratory staff of Environmental Engineering Laboratory, Department of Civil of Engineering, University of Moratuwa; Ms. Nilanthi Gunathilake Mr. Kasun Zoysa, Mr. Justin and Mr.Dhananjaya and Ms. Nipuni for the assistance received to conduct my research experiments successfully in the Environmental engineering laboratory.

I would like to thank the head and the staff of Analytical Laboratory, Department of Materials Science and Engineering, University of Moratuwa for allowing me to use laboratory equipment.

I would like to thank Madhurangi, Madhusha and Gimhani and Thilini for their friendship, guidance, strength, and assistance given in the period of the research study.

I am grateful to my family for being there for me, giving their unconditional love and support to fulfill my aims. Finally, I would like to thank my spouse and baby for supporting, encouraging, and understanding me in my quest.

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LIST OF ABBREVIATIONS

Abbreviation Description

CKD Chronic Kidney Disease

CKDu Chronic Kidney Disease of unknown etiology

CFA Coal Fly Ash

MFA Modified Fly Ash

FMFA Iron oxide Functionalized modified Fly Ash

(FMFA) opt Optimized Iron oxide Functionalized modified Fly Ash

IC Ion Chromatography

GFR Glomerular filtration rate

SL Sri Lanka

NCP North Central Province

FTIR Fourier-transform infrared spectroscopy

SEM Scanning electron microscope

EDX Energy-dispersive X-ray spectroscopy

XRD X-ray Diffraction Analysis