Removal of nutrients (N and P) and heavy metals (Fe, Al, Mn and Ni) from industrial wastewaters by phytoremediation using water hyacinth (*Eichhornia crassipes*) under different nutritional conditions

By

J.C. KASTURIARACHCHI



THIS THESIS WAS SUBMITTED TO THE DEPARTMENT OF CIVIL ENGINEERING OF THE UNIVERSITY OF MORATUWA IN PARTIAL FULFILLMENT OF THE REQUIREMENT FOR THE DEGREE OF MASTER OF SCIENCE.

DEPARTMENT OF CIVIL ENGINEERING UNIVERSITY OF MORATUWA SRI LANKA

624 03 628.316.12

SEPTEMBER 2003

82544



82544

The sis

DECLARATION

I certify that this dissertation does not incorporate without acknowledgement of any material previously submitted for a Degree or Diploma in any University and to the best of my knowledge and belief it does not contain any material previously published or written or orally communicated by another person except, where due references is made in the text.

J.C. Kasturiarachchi

MSc/C/22/2002

Certified by

UOM Verified Signature

University of Moratuwa, Sri Lanka. Electronic Theses & Dissertations www.lib.mrt.ac.lk

Dr MW. Jayaweera Supervisor

ABSTRACT

This study was investigated the utilization of phytoremediation strategies to remove nitrogen, phosphorus and heavy metals (Fe, Al, Mn and Ni) from wastewaters by water hyacinth (*Eichhornia crassipes* [Mart.] Solms). Batch studies were conducted for 15 weeks using fiberglass tanks in which healthy young plants were grown for a period of 15 weeks under different nutrient concentrations of 2-fold (56 TN mg/l and 15.4 TP mg/l), 1-fold (28 TN mg/l and 7.7 TP mg/l), 1/2-fold (14 TN mg/l and 3.85 TP mg/l), 1/4-fold (7 TN mg/l and 1.93 TP mg/l), 1/8-fold (3.5 TN mg/l and 0.96 TP mg/l) and control (without nutrients). In each week plants, water and sediments were analysed for TN and TP. The phytoremediation potential of heavy metal removal was determined at above nutrient concentrations with the addition of the constant heavy metal concentrations (Fe-9.27 mg/l, Al-5.62 mg/l, Mn-0.92 mg/l and Ni-0.21 mg/l) in fiberglass tanks. Plant, water and sediment were analyzed for heavy metals during the 15 weeks of culture period. A mass balance was conducted to investigate the phytoremediation efficiencies and to determine the different mechanisms governing nutrient and heavy metal removal from the wastewaters.

Our results manifested that hyacinth could be effectively utilized in constructed wetlands to phytoremediate N rich wastewaters than P. Plant uptake was the major TN and TP removal mechanism during the initial periods. Accumulation of a high content of nitrogen in plant tissues due to plant uptake and denitrification was found to be the key mechanisms involved in the efficient removal of nitrogen at the latter part of the study period. Plant uptake of phosphorus and chemical precipitation together with adsorption on to the detritus are the key mechanisms of phosphorus removal. However the phosphorus removal seems to be not high with that of nitrogen indicating that hyacinth systems are not ideal for phosphorus removal from wastewaters. In conclusion, very young plants having seems to be ideal to commence a constructed wetland after a period of acclimatization and approximately 56-63 days of hydraulic retention time is recommended for optimum phytoremediation of nitrogen as well as phosphorus.

Phytoremediation of Fe largely due to the process of rhizofiltration and the chemical precipitation followed by flocculation and sedimentation were the key Fe removal mechanisms during the first few weeks of the study. Plants grown in the control set-up showed a highest phytoremediation efficiency of 47% during optimum growth at the 6th week with a highest accumulation of 6707 Fe mg/kg dry weight. Root effluxing of Fe to the wastewaters at intermittent periods and with time was a key mechanism of avoiding Fe phytotoxicity in water hyacinth. It can be concluded from this study that water hyacinth is an ideal plant for a batch removal of low polluting Fe rich industrial wastewaters under completely nutrient poor conditions. Very young plants are ideal to commence a constructed wetland after a period of acclimatization and approximately 42 days hydraulic retention time is recommended for optimum phytoremediation. Phytoextraction was the key Mn removal mechanism and root effluxing of Mn was observed intermittently possibly to avoid any phytotoxicity caused by an excessive accumulation of Mn in hyacinths. Hyacinths cultured in the 1/8-fold set-up showed a highest accumulation of 1133 Mn mg/kg dry weight with an optimum removal of 79% at the 9th week. Hence very young plants inhabiting waterbodies containing approximately 3.5 TN mg/l and 0.96 TP mg/l seems to be more ideal for a batch removal of low polluting Mn rich wastewaters in constructed wetlands.

Acclimatization of the plants is necessary for at least 1 week prior to the removal of Mn and then approximately 63 days hydraulic retention time is recommended to optimize phytoremediation. Chemical precipitation followed by flocculation and sedimentation with phytoremediation mainly due to rhizofiltration were the key Al removal mechanisms Control and 1/8-fold set-ups showed higher phytoremediation efficiencies of 63% and 54%, respectively with maximum accumulations of 4278 Al mg/kg dry weight and 4224 Al mg/kg dry weight, respectively. Therefore young plants of completely nutrient starved adult hyacinths seems to be more ideal for a batch removal of low polluting Al rich industrial wastewaters in pilot scale constructed wetlands. A hydraulic retention time of approximately 28 days is recommended for optimum removal after a period of acclimatization of the young plants. The results manifested that hyacinths are essentially Ni excluders since higher levels of Ni were detected in water throughout the study.





ACKNOWLEDGEMENT

I am very grateful to Dr M.W. Jayaweera, Senior Lecturer, Department of Civil Engineering, University of Moratuwa, Sri Lanka being the main supervisor for granting me all necessities and advising me in all respects.

My sincere thanks go to Dr S.L.J. Wijeyekoon and Dr W.K. Hirimburegama, for their valuable comments and suggestions on the research work and the manuscript of my publications as co-supervisors.

I am very much grateful and thankful to the SIDA (Swedish International Development Cooperation Agency) for funding my research project.

I wish to thank Miss Priyanka Dissanayake, Mrs Nilanthi Gunatilaka and Mr Justin of the Environmental Engineering Laboratory at the University of Moratuwa, Mr K. Premadasa at the Department of Plant Sciences, University of Colombo who helped me in making this endeavor a success.

My special thanks go to Miss P.U.D. Fernando and all my friends who helped me in various ways to complete this research.

I am eternally grateful to my parents and my brothers for their encouragement and giving support throughout the study. or Morenium Sri Lanka.

www.lib.mrt.ac.lk

TABLE OF CONTENTS

List of Tables List of Figures Abbreviations	i ii iv
Chapter 1: INTRODUCTION	
1.0. Introduction1.1. Sources of pollutants1.1.1. Nutrients (N and P)1.1.2. Heavy metals	1 1 1
1.2. Environmental and public health consideration1.2.1. Nutrients1.2.2. Heavy metals	3 3 2
1.3. Conventional methods to remove nutrients and heavy metals1.4. Phytoremediation and water hyacinth1.5. Objective of the research	6
Chapter 2: LITERATURE REVIEW	
 2.1. Phytoremediation 2.2. Categories of phytoremediation 2.3. Background of plants used for phytoremediation 2.4. Genetic engineering 2.5. Microorganisms assisted phytoremediation 2.6. Ways of optimization 	10 11 13 17 18 20
Chapter 3: METHODOLOGY	
 3.0. Methodology 3.1. Nutrient removal 3.1.1. Experimental set-up 3.1.2. Analysis of physical and chemical parameters 	21 21 21 22
3.2. Heavy metal removal 3.2.1. Experimental set-up 3.2.2. Preparation of plant matter for heavy metal analysis 3.2.3. Sampling of water and sediments 3.2.4. Analysis of heavy metals 3.2.5. Analysis of physical and chemical parameters 3.2.6. Growth parameters 3.2.7. SEM photographs	22 22 23 24 24 24 25 25

Chapter 4: **RESULTS AND DISCUSSION**

4.1. Nutrient removal		
4.1.1. Physicochemical parameters	26	
4.1.2. Biomass production and growth	27	
4.1.3. Nitrogen and phosphorus removal	32	
4.1.4. Design criteria for constructed wetland system with		
water hyacinth (batch reactor systems)	38	
4.2. Heavy metal removals	42	
4.2.1. Physicochemical parameters	42	
4.2.2. Mechanisms of Fe removal	48	
4.2.3. Mechanisms of Mn removal	60	
4.2.4. Mechanisms of Al removal	68	
4.2.5. Mechanisms of Ni removal	78	
Chapter 5: CONCLUSIONS		
5.1. Nutrient removal	86	
5.2. Heavy metal removal	87	
Appendix	90	
Bibliography	95	
University of Moratuwa, Sri Lanka. Electronic Theses & Dissertations www.lib.mrt.ae.lk		

LIST OF TABLES

Table	Page
Table 3.1: Mineral nutrient composition of the standard (1-fold) culture solution	21
Table 3.2: Heavy metal concentrations in artificial wastewaters	23
Table 4.1: Percentage removal of TN and TP by different key mechanisms in different time periods	33
Table 4.2: Area required for the removal of nitrogen (m ²)	38
Table 4.3: Area required for the removal of phosphorus (m ²)	38
Table 4.4: Fe accumulation in water hyacinth tissues	54
Table 4.5: Mn accumulation in water hyacinth tissues	63
Table 4.6: Al accumulation in water hyacinth tissues	71
Table 4.7: Ni accumulation in water hyacinth tissues University of Moratuwa, Sri Lanka	82

LIST OF FIGURES

Figure	Pa	age
	Schematic simplified diagram of nitrogen cycling in a water body containing water hyacinth	7
•	Schematic simplified diagram of phosphorus cycling in a water body containing water hyacinth	8
_	Schematic simplified diagram of heavy metal cycling in a water body containing water hyacinth	8
Figure 3.1:	Experimental tanks containing water hyacinth populations	22
Figure 4.1:	Temperature variation in the culture solutions	26
Figure 4.2:	Dissolved oxygen distribution in the culture media	27
Figure 4.3:	pH variation in the culture media	27
Figure 4.4:	Biomass production in different nutrient media	28
Figure 4.5:	Biomass distribution in 2-fold and 1/8-fold culture solutions	28
Figure 4.6:	Variation of biomass production during the study period	30
Figure 4.7:	Distribution of root: shoot ratio	30
Figure 4.8:	Variation of nitrogen removal rates in different culture solutions	32
Figure 4.9:	Variation of phosphorus removal rates in different culture solutions	32
Figure 4.10:	: TN variation in water hyacinth	36
Figure 4.11:	: TP variation in water hyacinth	36
Figure 4.12:	Distribution of TN and TP in different plant tissues	37
Figure 4.13:	pH variation in the culture media	42
Figure 4.14:	Temperature variation in the culture media	43
Figure 4.15:	Turbidity variation in the culture media	44
Figure 4.16:	Dissolved oxygen variation in the culture media	46
Figure 4.17:	Sulphate variation in the culture media	47

Figure 4.18: Fe distribution in different compartments of the system	49
Figure 4.19: Initial chemical precipitation (milky-white)	50
Figure 4.20: Heavy metal adsorption to root surface	51
Figure 4.21: Fe distribution in above ground and below ground tissues	53
Figure 4.22: Fe bioaccumulation coefficient	59
Figure 4.23: Mn distribution in different compartments of the system	61
Figure 4.24: Mn distribution in above ground and below ground tissues	64
Figure 4.25: Mn bioaccumulation coefficient	67
Figure 4.26: Al distribution in different compartments of the system	69
Figure 4.27: Al distribution in above ground and below ground tissues	72
Figure 4.28: Al bioaccumulation coefficient	77
Figure 4.29: Ni distribution in different compartments of the system	80
Figure 4.30: Ni distribution in above ground and below ground tissues	81
Figure 4.31: Ni bioaccumulation coefficient	83

ABBREVIATIONS

AAS -Atomic Absorption Spectrophotometer 1-Aminocyclopropane-1-carboxylic acid ACC -

AS Aluminosilicate

BAC -Bioaccumulation coefficient BOD -Biochemical Oxygen Demand Cation Exchange Capacity CEC -Deoxyribonucleic Acid DNA -Dissolved Oxygen DO

Hydraulic Retention Time HRT

Iron Transporter ITRI

MAO -Monoamine Oxidase Activities

MT Metallothioneins

PAHs -Polycyclic Aromatic Hydrocarbons

PC **Phytochelatins**

Poly Chlorinated Biphenyl PCB

Pentachlorophenol PCP Standard Deviation SD

Subsurface Flow Wetlands SFW -Sediment oxygen Demand SOD -SRB Sulphate Reducing Bacteria

TCE

Trichloroethylene Total Nitrogen schools these & Dissertations TN

Trinitrotoluene www.lib.mrt.ac.lk TNT

Total Phosphorus TP

TSS Total Suspended Solids